Chloroprene as a Source of Fine Chemicals: Palladium-Catalyzed Synthesis of Terminal Allenes

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Supporting Data

General. All anaerobic and/or moisture sensitive manipulations were carried out with standard Schlenk techniques under predried nitrogen or with glovebox techniques under prepurified argon. Tetrahydrofuran was distilled from sodium benzophenone-ketyl under nitrogen prior to use. Dpbp, 1 DPEphos, 2 [PdCl(η^3 -C₃H₅)]₂, 3 Pd₂(dba)₃·CHCl₃, 4 and PhCH=CCl₂⁵ were prepared by reported methods. (Z)-2-Chloro-1-phenyl-1,3-butadiene 5 was prepared from PhCH=CCl₂ and CH₂=CHMgBr in 79% isolated yield according to a procedure of Minato and Tamao⁶ and its NMR data were consistent with those reported previously. NaOMe, KO^tBu, MeCH(COOEt)(COMe), and CH2(COOMe)2 were purchased from Wako Pure Chemical Industries and used as received. MeCH(COOMe)₂ and (CH₂=CHCH₂)CH(COOMe)₂ were purchased from Aldrich Chemical Co. and used as received. HN(COOtBu)2 was purchased from Tokyo Chemical Industry and used without further purification. Chloroprene (50% v/v in toluene with 0.5% of catecol) was a generous gift from Denki Kagaku Kogyo Co. Ltd. and used after distillation. The distilled chloroprene contained ca. 10 mol% of toluene (checked by NMR) but used for the following reactions without further purification. Reaction progress was monitored by analytical TLC using 0.25 mm Merck F-254 silica gel glass plates. NMR spectra were recorded on a JEOL JNM LA500 spectrometer (¹H, 500 MHz; ¹³C, 125 MHz). ¹H and ¹³C chemical shifts are reported in ppm downfield of internal tetramethylsilane.

Palladium-Catalyzed Synthesis of Allenes. The reaction conditions and results are summarized in Table 1. A typical procedure is given for preparation of **3a**. The characterization data of the products are listed below.

Dimethyl 2-(buta-2,3-dienyl)-2-methylpropane-1,3-dioate (**3a**). To a mixture of $Pd_2(dba)_3$ ·CHCl₃ (161 mg, 156 μmol), DPEphos (170 mg, 316 μmol), MeCH(COOMe)₂ (**2a**, 850 mg, 5.82 mmol), and NaOMe (345 mg, 639 mmol) in THF (10 mL) was added chloroprene **1** (with ca. 10% of toluene; ca. 3.0 g, ca. 30 mmol) by means of syringe under nitrogen. The mixture was warmed to 70 °C (bath temp.) and refluxed for 3 h. The mixture was diluted with hexane (ca. 100 mL) and the precipitate was removed by filtration. The mother liquor was concentrated using a rotary evaporator and the residue was passed through a short pad of silica gel using ether as eluent. Concentration of the solution followed by vacuum transfer of the residue gave the terminal allene **3a** as colorless liquid. Yield: 1.08 g (5.45 mmol, 94%). ¹H NMR (CDCl₃): δ 1.44 (s, 3H), 2.58 (dt, J = 7.9 and 2.4 Hz, 2H), 3.73 (s, 6H), 4.67 (dt, J = 6.7 and 2.4 Hz, 2H), 5.00 (tt, J = 7.9 and 6.7 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 19.78, 35.27, 52.52, 53.97, 74.55, 84.51, 172.19, 210.22. Anal. Calcd for $C_{10}H_{14}O_4$: C, 60.59; H, 7.12. Found: C, 60.62; H, 7.13.

Dimethyl 2-(buta-2,3-dienyl)propane-1,3-dioate (**3b).** ¹H NMR (CDCl₃): δ 2.60 (ddt, J = 7.4, 6.7, and 3.1 Hz, 2H), 3.51 (t, J = 7.4 Hz, 1H), 3.74 (s, 6H), 4.72 (dt, J = 7.4)

6.7 and 3.1 Hz, 2H), 5.14 (quint, J = 6.7 Hz, 1H). $^{13}C\{^{1}H\}$ NMR (CDCl₃): δ 27.40, 51.27, 52.53, 76.23, 86.63, 169.24, 208.72. Anal. Calcd for $C_{9}H_{12}O_{4}$: C, 58.69; H, 6.57. Found: C, 58.78; H, 6.86.

Dimethyl 2,2-bis(buta-2,3-dienyl)propane-1,3-dioate (**3b').** ¹H NMR (CDCl₃): δ 2.65 (dt, J = 8.1 and 2.4 Hz, 4H), 3.73 (s, 6H), 4.67 (dt, J = 6.6 and 2.4 Hz, 4H), 4.94 (tt, J = 8.1 and 6.6 Hz, 2H). ¹³C{¹H} NMR (CDCl₃): δ 31.96, 52.47, 57.91, 74.66, 84.10, 170.94, 210.13. Anal. Calcd for C₁₃H₁₆O₄: C, 66.09; H, 6.83. Found: C, 65.98; H, 6.82.

Dimethyl 2-(2-propenyl)-2-(buta-2,3-dienyl)propane-1,3-dioate (**3c**). 1 H NMR (CDCl₃): δ 2.61 (dt, J = 7.9 and 2.4 Hz, 2H), 2.69 (dt, J = 7.4 and 1.1 Hz, 2H), 3.72 (s, 6H), 4.66 (dt, J = 6.6 and 2.4 Hz, 2H), 4.95 (tt, J = 7.9 and 6.6 Hz, 1H), 5.09-5.14 (m, 2H), 5.65 (ddt, J = 16.7, 10.3, and 7.4 Hz, 1H). 13 C{ 1 H} NMR (CDCl₃): δ 31.94, 36.90, 52.39, 57.93, 74.63, 84.17, 119.33, 132.27, 171.04, 210.15. Anal. Calcd for C₁₂H₁₆O₄: C, 64.27; H, 7.19. Found: C, 64.32; H, 7.17.

Ethyl 2-acetyl-2-methylhexa-4,5-dienoate (**3d**). ¹H NMR (CDCl₃): δ 1.27 (t, J = 7.1 Hz, 3H), 1.37 (s, 3H), 2.16 (s, 3H), 2.49 (ddt, J = 14.2, 8.2, and 2.4 Hz, 1H), 2.60 (ddt, J = 14.2, 7.6, and 2.8 Hz), 4.19 (dq, J = 10.8 and 7.1 Hz, 1H), 4.21 (dq, J = 10.8 and 7.1 Hz, 1H), 4.66 (ddd, J = 6.7, 2.8, and 2.4 Hz, 2H), 4.97 (ddt, J = 8.2, 7.6, and 6.7 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 14.09, 18.89, 26.29, 34.42, 59.79, 61.43, 74.68, 84.63, 172.41, 205.00, 210.06. Anal. Calcd for C₁₁H₁₆O₃: C, 67.32; H, 8.22. Found: C, 67.05; H, 8.06.

Di-*tert*-butyl *N*-(buta-2,3-dienyl)iminodicarboxylate (3e). ¹H NMR (CDCl₃): δ 1.51 (s, 18H), 4.18 (dt, J = 6.2 and 2.9 Hz, 2H), 4.78 (dt, J = 6.6 and 2.9 Hz, 1H), 5.21 (tt, J = 6.6 and 6.2 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 28.12, 44.73, 76.64, 82.35, 87.58, 152.21, 208.82. Anal. Calcd for C₁₄H₂₃O₄N: C, 62.43; H, 8.61; N, 5.20. Found: C, 62.28; H, 8.37; N, 5.12.

tert-Butyl 2-(tert-butoxycarbonylamino)hexa-4,5-dienoate (3f). ¹H NMR (CDCl₃): δ 1.45 (s, 9H), 1.47 (s, 9H), 2.44 (br, 1H), 2.51 (br, 1H), 4.28 (br, 1H), 4.70 (d, J = 5.6 Hz, 2H), 5.01 (m, 1H), 5.13 (br d, J = 7.0 Hz, 1H). ¹³C{¹H} NMR (CDCl₃): δ 28.02, 28.36, 32.15, 53.62, 75.12, 79.69, 82.01, 84.59, 155.19, 170.96, 209.68. Anal. Calcd for C₁₅H₂₅O₄N: C, 63.58; H, 8.89; N, 4.94. Found: C, 63.54; H, 8.83; N, 4.83.

References

- (1) Ogasawara, M.; Yoshida, K.; Hayashi, T. Organometallics 2000, 19, 1567.
- (2) Kranenburg, M.; van der Burgt, Y. E. M.; Kamer, P. C. J.; van Leeuwen, P. W. N. M.; Goubitz, K.; Fraanje, J. *Organometallics* **1995**, *14*, 3081.
- (3) Tatsuno, A.; Yoshida, T.; Otsuka, S. *Inorg. Synth.* **1979**, *19*, 220.
- (4) Ukai, T.; Kawazura, H.; Ishii, Y.; Bonnet, J. J.; Ibers, J. A. J. Organomet. Chem. 1974, 65, 253.
- (5) Rabinowitz, R.; Marcus, R. J. Am. Chem. Soc. 1962, 84, 1312.
- (6) Minato, A.; Suzuki, K.; Tamao, K. J. Am. Chem. Soc. 1987, 109, 1257.
- (7) Krijnen, E. S.; Zuihof, H.; Lodder, G. J. Org. Chem. 1994, 59, 8139.